

# ELEMENTS COMPOUNDS AND MIXTURES



Types: \_\_\_\_\_ Date: \_\_\_\_\_

## Elements and Compounds

### Mixtures vs. Compounds

Directions: Complete whether each mixture described is uniform or nonuniform. Write "U" for uniform and "N" for nonuniform.

1. Air	Answer: _____
2. Tap water	Answer: _____
3. Fruit salad	Answer: _____
4. Sugar	Answer: _____
5. Lemonade	Answer: _____
6. Salt water	Answer: _____
7. Carbon dioxide (CO <sub>2</sub> )	Answer: _____
8. Pure gold	Answer: _____
9. Pure iron	Answer: _____
10. Pure copper	Answer: _____

Name: \_\_\_\_\_ Class: \_\_\_\_\_

## Elements and Compounds

Directions: (Identify whether each given item is an element or a compound. If it is an element, write down its chemical symbol. If it is a compound, write down its chemical formula.)

Ex. Oxygen

Type	Chemical Symbol
Element	O

1. Water

Type	Chemical Symbol

2. Carbon Dioxide

Type	Chemical Symbol

Name: \_\_\_\_\_

## Elements and Compounds

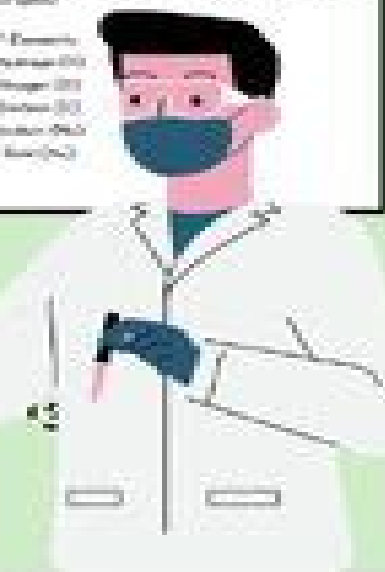
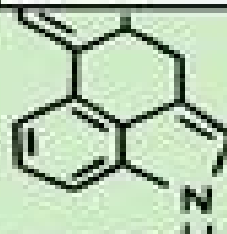
### Element

Directions: (Define: A pure substance made up of only one type of atom. Examples: Hydrogen (H<sub>2</sub>), Oxygen (O<sub>2</sub>), Carbon (C), Gold (Au).)

Directions: (Define: A substance composed of two or more different types of atoms. Examples: Water (H<sub>2</sub>O), Carbon dioxide (CO<sub>2</sub>), Air (mixture).)

Examples of Elements:

- Hydrogen (H<sub>2</sub>)
- Oxygen (O<sub>2</sub>)
- Carbon (C)
- Gold (Au)
- Silver (Ag)



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# Elements And Compounds And Mixture Study Guide

**Morris Hein,Susan Arena**



## Elements And Compounds And Mixture Study Guide:

*Chemistry, Student Study Guide* James E. Brady, Fred Senese, 2008-01-28 The image on the front cover depicts a carbon nanotube emerging from a glowing plasma of hydrogen and carbon as it forms around particles of a metal catalyst Carbon nanotubes are a recently discovered allotrope of carbon Three other allotropes of carbon buckyballs graphite and diamond are illustrated at the left as is the molecule methane  $\text{CH}_4$  from which nanotubes and buckyballs can be made The element carbon forms an amazing number of compounds with structures that follow from simple methane found in natural gas to the complex macromolecules that serve as the basis of life on our planet The study of chemistry also follows from the simple to the more complex and the strength of this text is that it enables students with varied backgrounds to proceed together to significant levels of achievement

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