

Name _____

Elements, Compounds & Mixtures Worksheet

Part 1: Read the following information on elements, compounds and mixtures. Fill in the blanks where necessary.

Elements:

- A pure substance containing only one kind of atom.
- An element is always uniform all the way through (homogeneous).
- An element cannot be separated into simpler materials (except during nuclear reactions).
- Over 100 existing elements are listed and classified on the Periodic Table.

Compounds:

- A pure substance containing two or more kinds of atoms.
- The atoms are chemically combined in some way. Often times (but not always) they come together to form groups of atoms called molecules.
- A compound is always homogeneous (uniform).
- Compounds cannot be separated by physical means. Separating a compound requires a chemical reaction.
- The properties of a compound are usually different than the properties of the elements it contains.

Mixtures:

- Two or more elements or compounds NOT chemically combined.
- No reaction between substances.
- Mixtures can be uniform (called homogeneous) and are known as solutions.
- Mixtures can also be non-uniform (called heterogeneous).
- Mixtures can be separated into their components by chemical or physical means.
- The properties of a mixture are similar to the properties of its components.

Part 2: Classify each of the following as elements (E), compounds (C) or Mixtures (M). Write the letter X if it is none of these.

<u>E</u> Diamond	<u>C</u> Sugar ($C_6H_{12}O_6$)	<u>M</u> Milk	<u>E</u> Iron (Fe)
<u>M</u> Air	<u>C</u> Sulfuric Acid (H_2SO_4)	<u>M</u> Gasoline	<u>X</u> Electricity
<u>E</u> Krypton (Kr)	<u>E</u> Bismuth (Bi)	<u>E</u> Uranium Dioxide	<u>M</u> Popcorn
<u>C</u> Water (H_2O)	<u>C</u> Alcohol (CH_3OH)	<u>M</u> Pail of Garbage	<u>M</u> A dog
<u>C</u> Ammonia (NH_3)	<u>C</u> Salt ($NaCl$)	<u>X</u> Energy	<u>E</u> Gold (Au)
<u>M</u> Wood	<u>M</u> Bronze	<u>M</u> Ink	<u>M</u> Pizza
<u>C</u> Dry Ice (CO_2)	<u>C</u> Baking Soda ($NaHCO_3$)	<u>E</u> Titanium (Ti)	<u>M</u> Concrete

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